

De-Convolution of Complex Residual Gas Spectra at JET

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*See annex of F Romanelli et al, "Overview of JET Results", (Proc. 22nd IAEA Fusion Energy Conference, Geneva, Switzerland (2008))

INTRODUCTION

At JET, fusion fuel most commonly used in the tokamak is deuterium. Purity of the fuel is vital to ensure the data obtained from plasma pulses is useful, and that pulse recipe repeatability is not compromised.

Our primary goal was to determine the purity of the fuelling gas, deuterium, and confirm that we did not have any contamination from helium (He). Analysis using our quadrupole Residual Gas Analysers (qRGA) in the conventional way was not possible due to the overlapping masses of the two species at 4 atomic mass units (amu).

Here, we present and demonstrate the approach we used to solve this problem; a complimentary technique, Threshold Ionization Mass Spectrometry (TIMS), operating the qRGA mass spectrometer in a mode allowing control over the energy of the electrons emitted within the ionization source.

Recognising the potential to identify more complex compounds which have brought uncertainty when attempting to interpret data from the torus primary vacuum vessel we have applied this technique across a broad range of gases that exhibit overlapping masses in the laboratory.

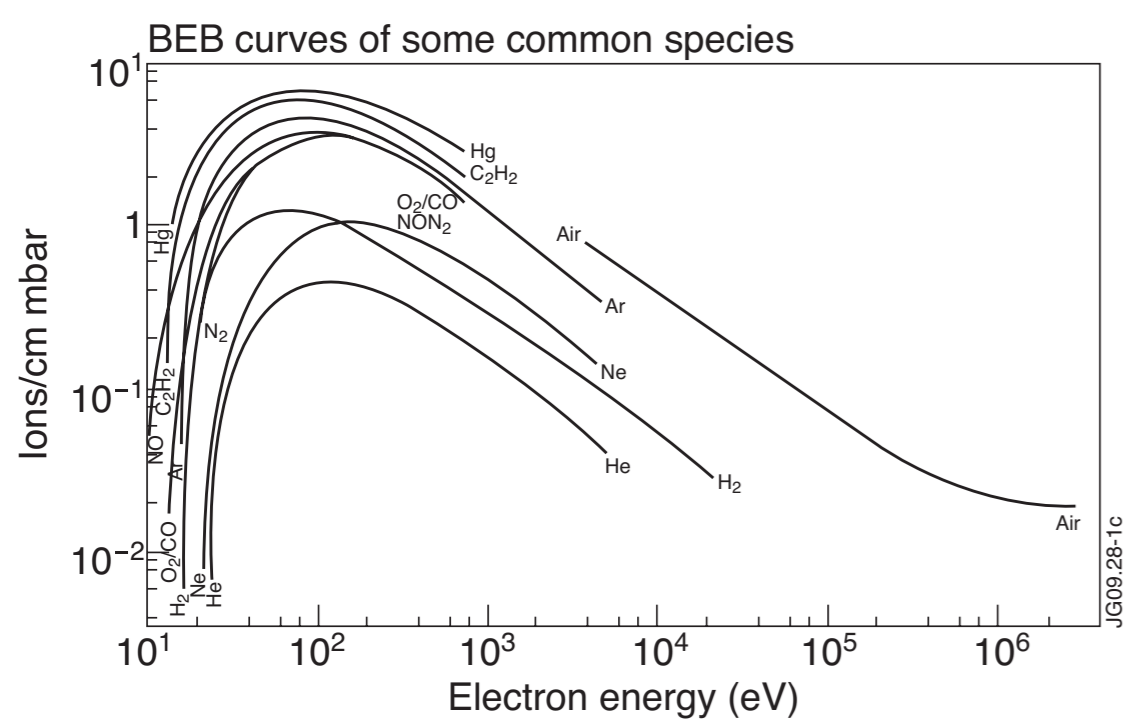
JET is unique due to the many interacting species.

- 3 isotopes of hydrogen isotope: hydrogen, deuterium and tritium.
- 2 isotopes of helium: ³He, ⁴He
- 3 isotopes of water, H₂O/ heavy water, D₂O / super heavy water, T₂O.
- 6 Tonnes of carbon tiles, complex carbon / hydrogen isotope interactions.

Baking the vessel to 320°C further complicates an already difficult task due to dissociation of C_xH_y and C_xD_y species from the vessel wall.

Experimental Details of TIMS at JET

The Hiden Analytical qRGA system at JET is operated in a mode where the energy of the electrons emitted within the ionization source is variable. Different elements have defined ionization energies required to remove an orbiting electron. This energy is dependent on the electron orbital, i.e. outer shell electrons generally have weaker ionization energies due to the greater distance and lower electrostatic forces from the nucleus. This gives rise to the electron impact "threshold ionization energy" curve. The ionization process of neutral particles commences at a minimum (threshold) energy of the impacting electrons. This minimum energy is dependent and unique to any species present in the gas matrix, resulting in a spectral "identifier" or fingerprint for all atomic or molecular species.



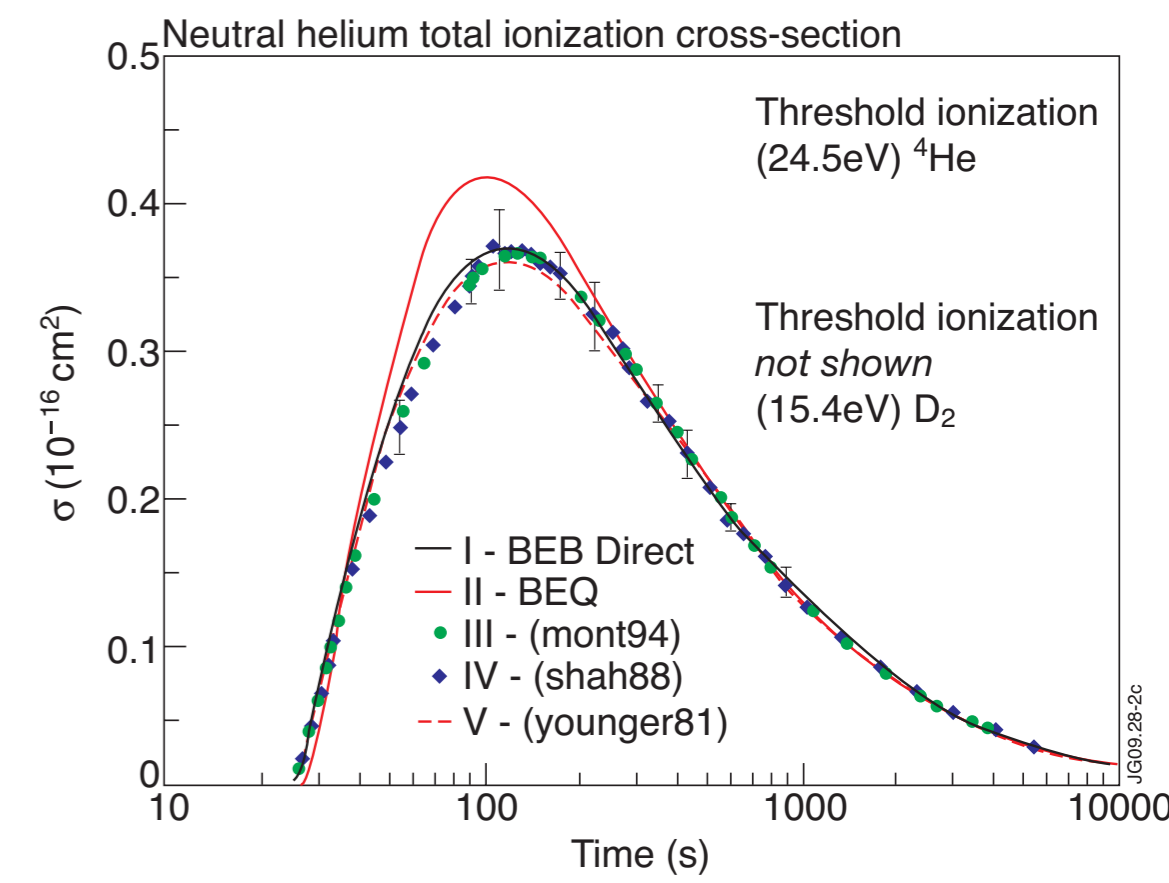
Halzoric and Test chamber

Experimental Method

- Hiden Analytical HAL201 RC qRGA mass spectrometers interfaced to test chamber at JET. Total system pressure held at $\sim 10^{-7}$ mbar during data collection.
- Gas mixtures introduced through the test chamber with Mass Flow Controllers (MFC) to allow user control of the ratio of individual species within the matrix. (MFC accuracy $\pm 1\%$).
- Gas species in the test chamber (mixed using the MFC's); ultra pure (100%) D₂, (100%) ⁴He, and small (1%) concentrations of CH₄, C₂H₄, C₂H₆, C₃H₆, C₃H₈, C₄H₁₀ in (94%) N₂ carrier gas.

Experimental Aims

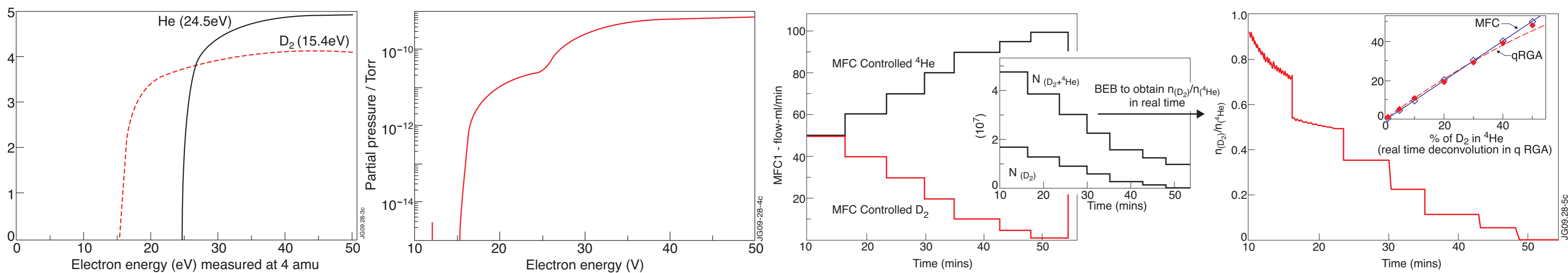
- Mix D₂ and ⁴He and control ratio of the mixture with MFC's. Investigate the possibility of de-convolution of D₂ from ⁴He at 4amu in the mass spectrum, and real-time quantification analysis techniques using the qRGA TIMS technique, combined with BEB ionization cross section (σ) theory.
- Mix D₂ with the hydro-carbon mixture to investigate the possibility of TIMS de-convolution of complex H_xD_y, C_xH_y, C_xD_y, and preliminary real time qualitative data analysis. Investigate D₂O formation and de-convolution at 20amu



Gas inlet system

Results: TIMS De-convolution of ⁴He/D₂ Gas Mixture at JET

The figure below shows real time mass flow controlled ratio of ⁴He/D₂ gas mixture displayed within qRGA software. The inset on the left shows the corresponding raw qRGA data [at 4amu] taken as a function of time. The two traces represent N(D₂) and N(D₂ + He) at 19eV and 31eV respectively. The real time qRGA data on the right shows the true ⁴He/D₂ abundance ratio n(D₂)/n(He) after applying BEB theory in the qRGA software.



Application of the BEB threshold ionization cross section theory to qRGA data:

From theory, N(D₂) is the D₂ partial pressure signal measured at 19eV and N(D₂ + He) is the D₂ and ⁴He partial pressure signal combined, measured at 31eV in the qRGA. Now, $\eta_{(D_2)}$ and $\eta_{(He)}$ are the real abundance densities of D₂ and ⁴He in the sample gas. To obtain the ratio, $n_{(D_2)}/n_{(He)}$ from the measured partial pressure signals N(D₂) and N(D₂ + He), to a first approximation it can be shown;

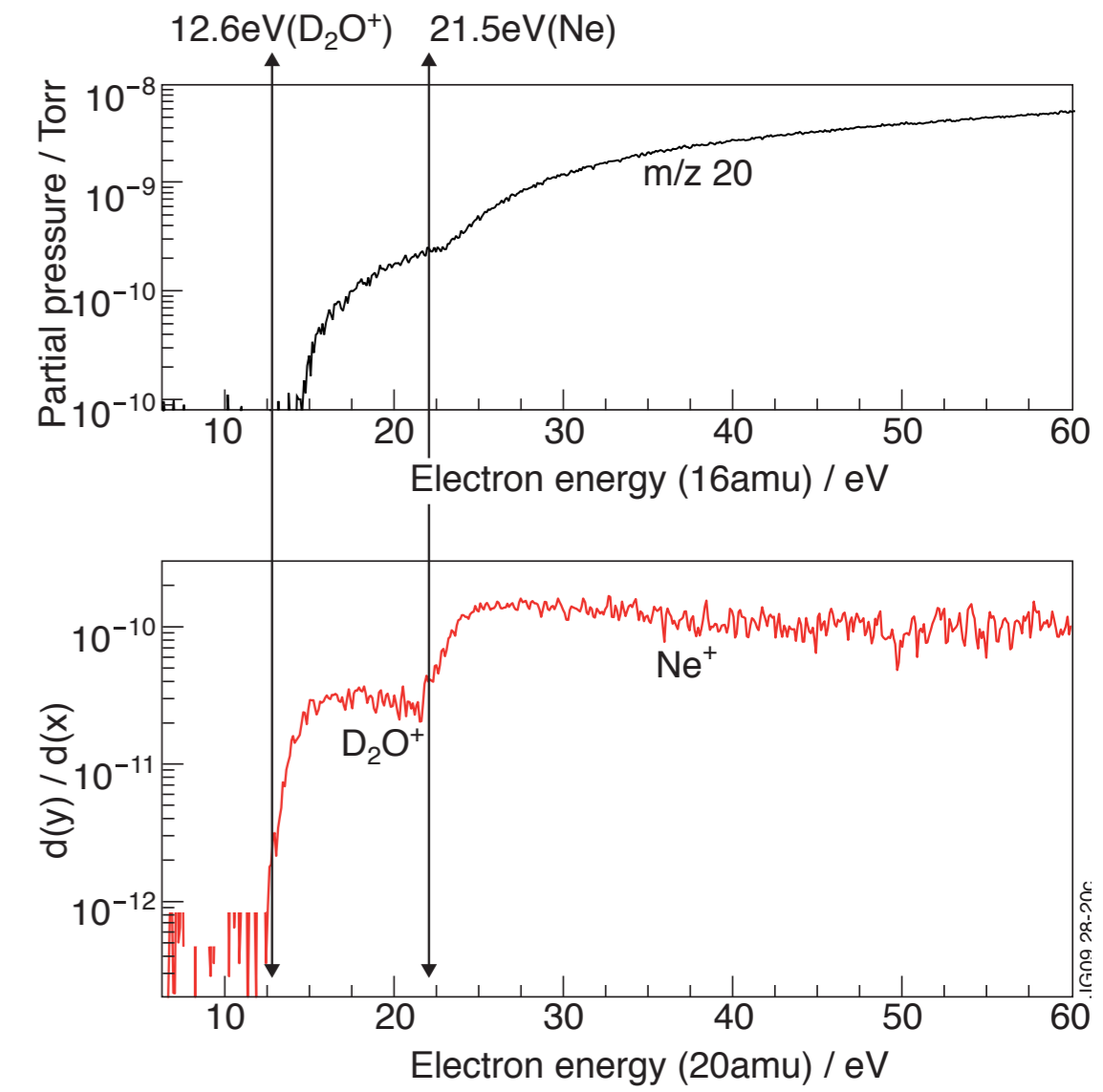
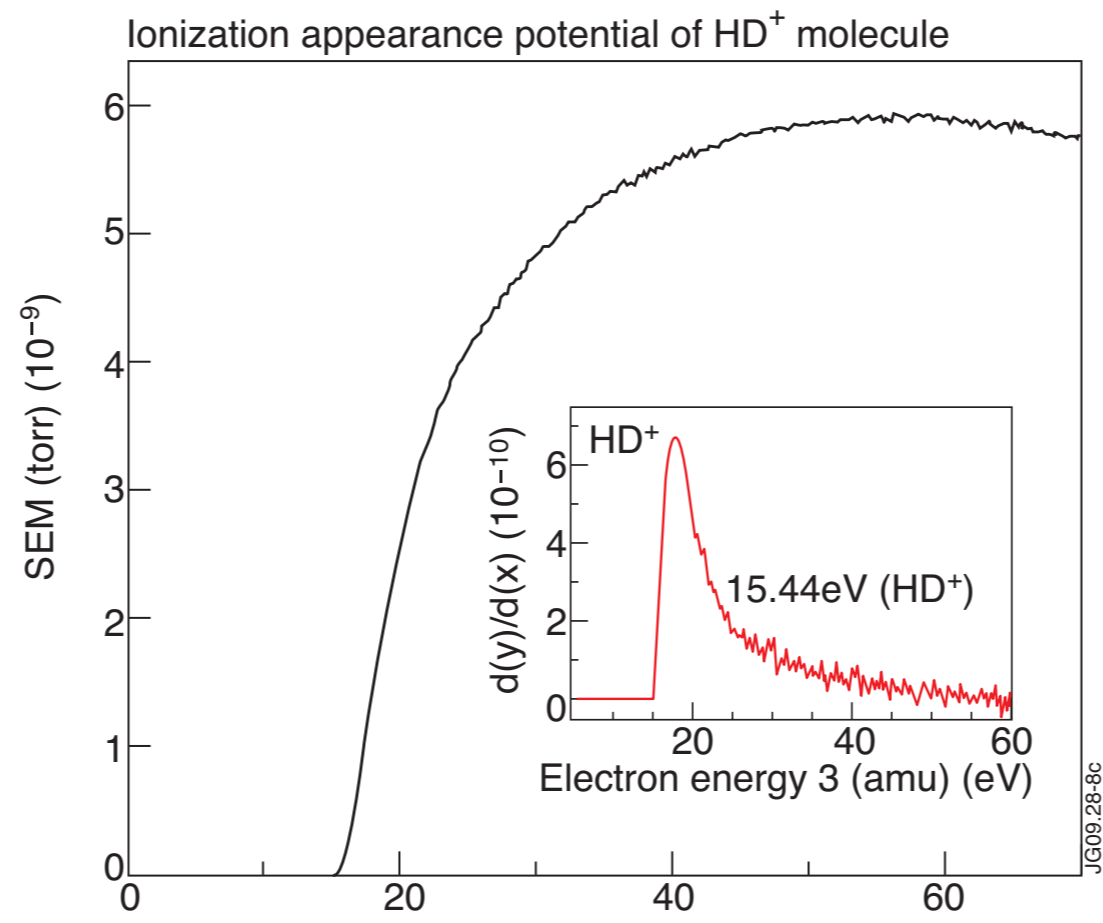
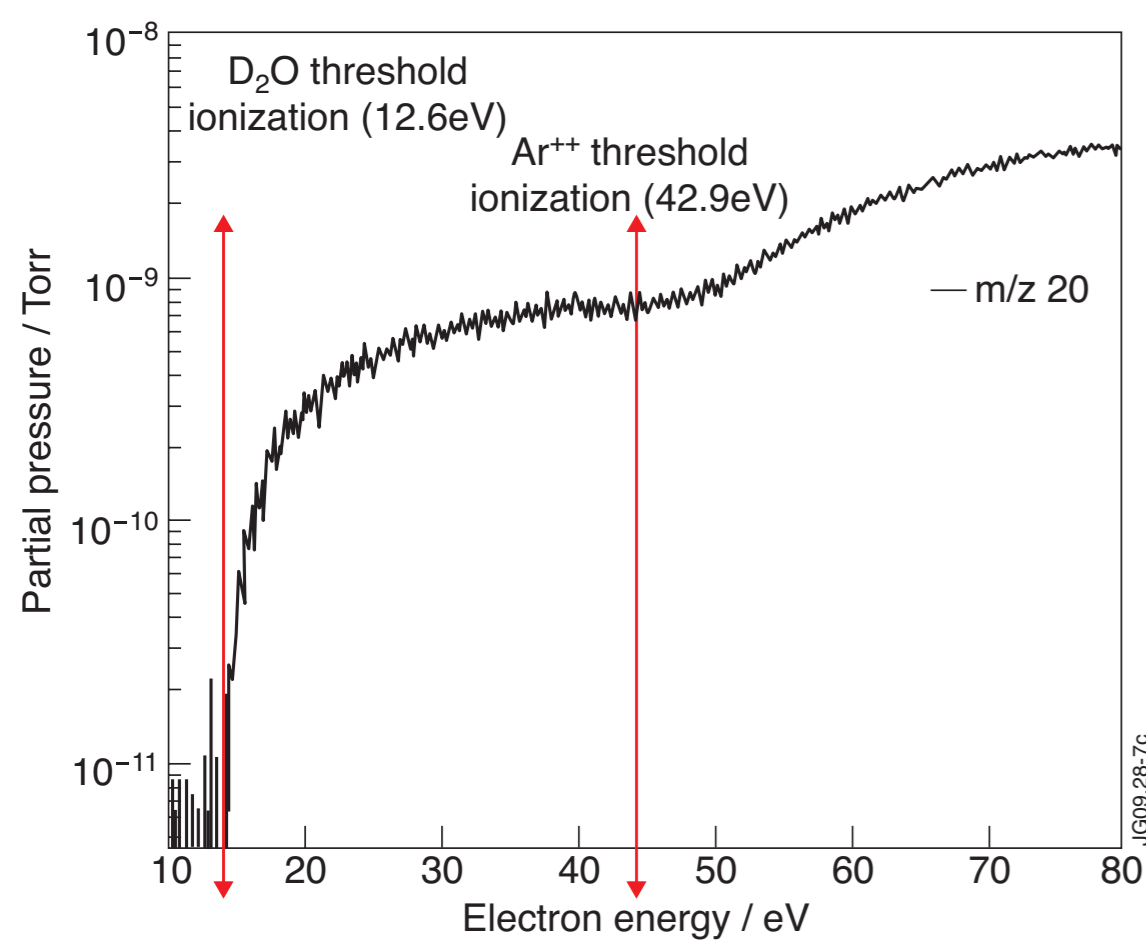
$$N_{(D_2)} = A_i \left(\sigma_{(D_2)}^{(19eV)} \right) \cdot \eta_{(D_2)} \quad (1), \quad \text{and similarly,} \quad N_{(D_2 + He)} = A_i \left\{ \left(\sigma_{(D_2)}^{(31eV)} \right) \cdot \eta_{(D_2)} + \left(\sigma_{(He)}^{(31eV)} \right) \cdot \eta_{(He)} \right\} \quad (2)$$

A_i is the instrument (qRGA) factor which is the same for both D₂ and ⁴He as the signal is measured at m/z = 4amu. It is also independent of electron ionization energy between energies of 19 and 31eV. Also, we define the ionization cross sectional coefficients for D₂ and He as $\eta_{(D_2)}$ and $\eta_{(He)}$ respectively: Solving (1) and (2) yields;

$$\frac{\eta_{(D_2)}}{\eta_{(He)}} = \frac{\left(\sigma_{(He)}^{(31eV)} \right)}{\frac{N_{(D_2 + He)}}{N_{(D_2)}} \left(\sigma_{(D_2)}^{(19eV)} \right) - \left(\sigma_{(D_2)}^{(31eV)} \right)}$$

TIMS de-convolution of $\leq 1\%$ heavy water (D_2O) from doubly ionized argon (Ar) at 20 amu.

The figure below shows separation of small ($<1\%$) quantities of heavy water from chamber residual Argon. Neon (Ne) at mass 20 has also been separated in this way which is a useful tool to identify leakages from the JET torus interspaces which are filled with this gas. It can also be seen that TIMS clearly identifies formation of HD^+ molecule at 3amu via the onset at 15.5eV. This will allow confident de-convolution from T^+ (3amu) when Tritium is introduced into the JET fuel mixture. (note the inset shows the derivative of the TIMS HD^+ curve to qualitatively further enhance species de-convolution in the TIMS spectra)

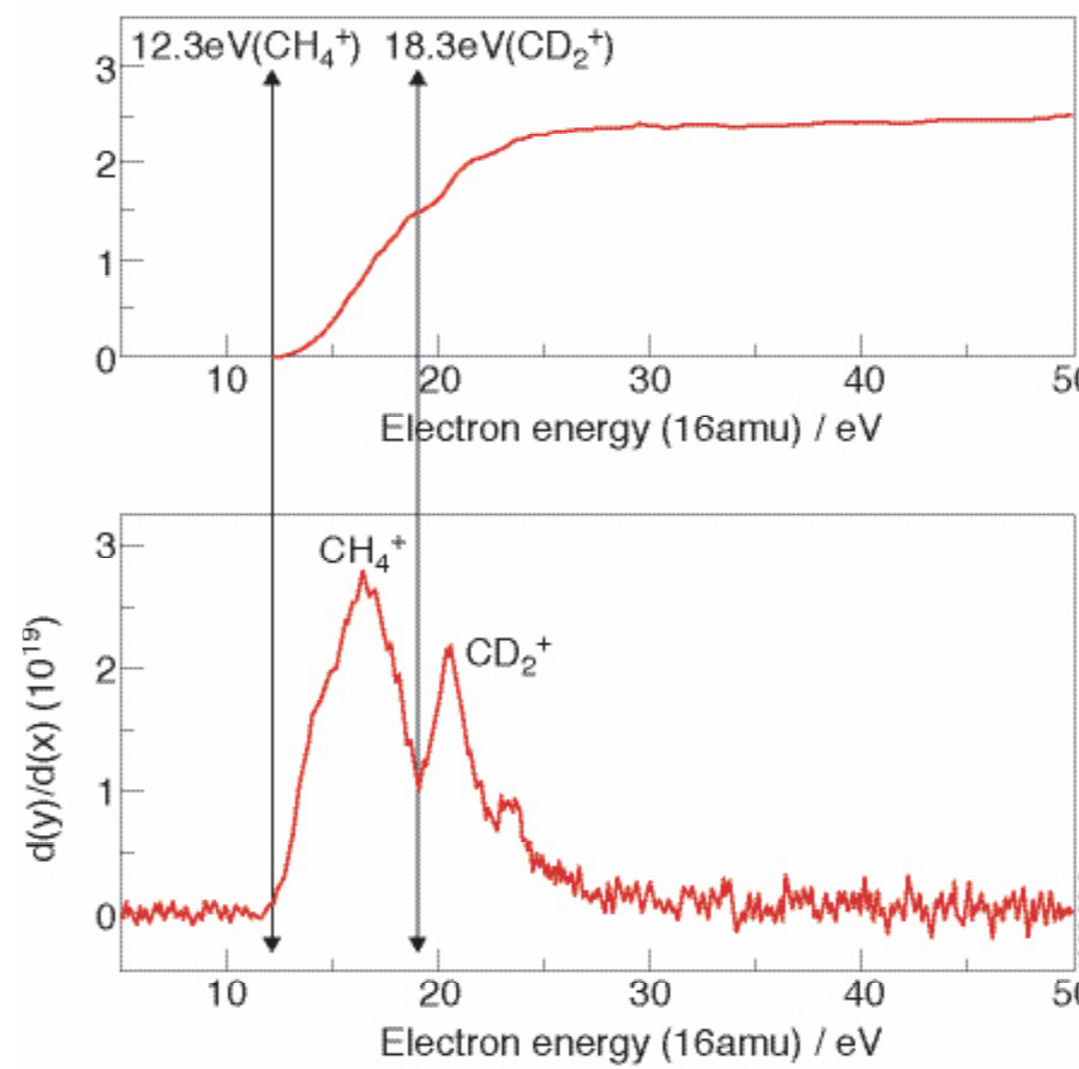


Preliminary results show interaction of Deuterium (D_2) and hydrocarbons resulting in the formation of C_xH_y and C_xD_y molecules. For example at 16amu, overlapping species are CH_4^+ and CD_2^+ . TIMS onsets are 12.3eV and 18.24eV respectively. Below species de-convolution is shown with TIMS. The differential;

$$\frac{dy}{dx} = \frac{d(\text{partial pressure})}{d(\text{electron energy})}, \text{ (dy/dx) is shown simply to}$$

qualitatively enhance the de-convolution of the transition regions, with the minima in the corresponding to the transition from CH_4^+ to $(CH_4^+ + CD_2^+)$

In the figure on the right, the hydro-carbons in the 50:50 $D_2/(N_2 \text{ carrier gas})$ are present at just a 0.5% fraction of the gas mixture total composition, also demonstrating a use for low level impurity detection.



Neon (Ne) at mass 20 has also been separated in this way. The two spectra above show the raw data curve (top) and the derivative (bottom) clearly demonstrating the separation of the D_2O at 12.6eV and Ne at 21.5eV, both sharing the same mass at 20amu. Separation of these masses will prove a useful tool at JET as the vessel interspaces are filled with neon so detection of a leak from one of these into the torus can be identified easily.

Whilst here we have demonstrated findings from some of our first laboratory experiments our success has identified characteristics of many of the species in this table. Occasionally overlapping species may have first ionization energies too close to each other to distinguish with our equipment. These cases are easily overcome but utilising the 2nd or 3rd ionisation energies by increasing the energy of the electrons emitted within the qRGA source to eject orbiting electrons from more tightly bound shells.

This table lists many species showing information of the respective masses that go to form the individual "cracking patterns". Many of the hydrogen and deuterium carbon species have several minor masses due to fragmentation of the molecules within the instrumentation electron bombardment source. As in this case cracking patterns are shown from instrumentation operating with a fixed electron energy of 70eV. The column on the right of the table indicated the 1st ionization potential for each respective species.

Conclusions and Future Work

Using the qRGA TIMS technique, de-convolution of the mass spectra obtained at JET could lead the way to providing a better understanding of the chemistry within the vacuum vessel, as well as providing invaluable diagnostic information during vessel conditioning.

Initial TIMS has proved encouraging, demonstrating discrimination of $D_2/{}^4He$ (4amu) and D_2O/Ar^{++} (20amu). De-convolution of C_xH_y and C_xD_y molecules in the mass spectrum is ongoing, with initial results de-convoluting CH_4^+ and CD_2^+ (16amu). With application of BEB to TIMS, we have demonstrated the generation of algorithms within the qRGA software to automatically discriminate different species quantitatively with overlapping peaks in the mass spectra.

Further experimental work will continue using our dedicated gas inlet with the addition of a carbon reaction cell to study the dissociation of species with temperature as well as generate C_xH_y and C_xD_y species to obtain further TIMS data for the purpose of improving the technique.

In addition to this the installation of the Hiden HAL201 RC qRGA is taking place on the JET Gas Introduction System Matrix and on the JET torus. From these instruments we hope to prove our de-convolution algorithms to give an accurate breakdown of the composition of the complex residual gas spectra at JET.

2010 brings the installation of the beryllium plasma facing wall at JET and an opportunity to accurately determine the vessel chemistry during commissioning. The results from this phase of JET operations and further applications of the technique including leak detection and cryogenic pump regeneration inventory determination could prove valuable for ITER commissioning and operations.

ACKNOWLEDGEMENTS

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References

- [1]. Coyne, T, RGAs Used At JET, **RGA-8**, Culham (2008)
- [2]. Yong, K, J. Res. Nat. Inst. Stand. Technol. **105**, 285. (2000)

CH_4^+/CO_2^+ Tims spectra

| Species | Major m/z | % | Major1 m/z | % | Major2 m/z | % | Major3 m/z | % | Major4 m/z | % | Major5 m/z | % | Major6 m/z | % | Rel Sens | 1st IE (eV) |
|---------|-----------|-----|------------|----|------------|----|------------|----|------------|----|------------|----|------------|------|----------|-------------|
| H2 | 2 | 100 | | | | | | | | | | | | | | 15.42 |
| He | 4 | 100 | | | | | | | | | | | | | 0.14 | 24.58 |
| D2 | 4 | 400 | | | | | | | | | | | | | | 15.46 |
| CH4 | 16 | 100 | 15 | 85 | 14 | 16 | 13 | 5 | 12 | 2 | 17 | 1 | | 1.6 | 12.61 | |
| H2O | 18 | 100 | 17 | 21 | 16 | 2 | | | | | | | | 0.9 | 12.61 | |
| D2O | 20 | 100 | 18 | 21 | 16 | 2 | | | | | | | | 0.9 | | |
| CD4 | 20 | 100 | 18 | 85 | 16 | 16 | 15 | 5 | 14 | 2 | | | | 1.6 | 12.65 | |
| Ne | 20 | 100 | 22 | 10 | | | | | | | | | | | | 21.56 |
| Air | 28 | 100 | 32 | 27 | 14 | 6 | 40 | 1 | | | | | | 1 | | |
| CO | 28 | 100 | 12 | 5 | 16 | 2 | | | | | | | | 1.05 | 14.01 | |
| C2H6 | 28 | 100 | 27 | 33 | 39 | 26 | 26 | 23 | 29 | 21 | 15 | 5 | | 2.6 | 11.57 | |
| C2D2 | 28 | 100 | 26 | 17 | 14 | 5 | 24 | 4 | 12 | 1 | | | | | 11.2 | |
| C3H8 | 28 | 100 | 26 | 59 | 27 | 41 | 44 | 28 | 43 | 23 | 39 | 19 | 41 | 13 | | 10.94 |
| C2H4 | 28 | 100 | 27 | 64 | 26 | 53 | 25 | 8 | 14 | 2 | | | | | | 10.51 |
| N2 | 28 | 100 | 14 | 15 | | | | | | | | | | | | 15.58 |
| C2D4 | 32 | 100 | 28 | 64 | 30 | 62 | 16 | 11 | 26 | 10 | 14 | 6 | 12 | 4 | | 10.54 |
| C2D6 | 32 | 100 | 30 | 24 | 36 | 20 | 28 | 18 | 34 | 16 | 18 | 4 | 16 | 4 | | 11.73 |
| O2 | 32 | 100 | 16 | 22 | | | | | | | | | | | | 12.06 |
| C3D8 | 34 | 100 | 32 | 60 | 30 | 30 | 50 | 22 | 42 | 16 | 46 | 12 | 28 | 7 | | 11.4 |
| Ar | 40 | 100 | 20 | 16 | | | | | | | | | | | 1.2 | 15.75 |
| C3H4 | 40 | 100 | 39 | 80 | 38 | 30 | 37 | 20 | 36 | 5 | 41 | 4 | | | | 10.36 |
| C3H6 | 42 | 100 | 39 | 73 | 44 | 70 | 27 | 39 | 40 | 29 | 38 | 20 | 37 | 13 | | 9.73 |
| C4H10 | 43 | 100 | 29 | 44 | 27 | 40 | 28 | 32 | 41 | 40 | 39 | 14 | 58 | 13 | | 10.53 |
| CO2 | 44 | 100 | 16 | 9 | 28 | 8 | 14 | 8 | 22 | 3 | | | | | 1.4 | 13.79 |
| C3D4 | 44 | 100 | 16 | 89 | 40 | 35 | 38 | 27 | 36 | 8 | 26 | 4 | | | | 10.37 |
| C3D6 | 46 | 100 | 42 | 74 | 48 | 70 | 30 | 39 | 44 | 30 | 40 | 20 | 38 | 13 | | 9.76 |
| C4D10 | 50 | 100 | 34 | 44 | 30 | 37 | 32 | 37 | 46 | 28 | 42 | 13 | 48 | 12 | | 10.5 |