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Quadrupole Mass Spectrometry Concepts

Mass Spectrometers for Residual Gas Analysis

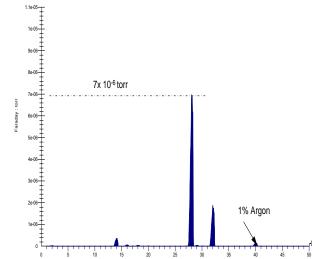






What does Residual Gas Analysis allow us to do?

- RGA is the examination of the molecular components present in a vessel or evolved from a system.
 It allows us to analyse, ON-LINE and in REAL time:
- Base Pressure Fingerprint
- Leak Detection
- Virtual Leaks / desorption
- Outgassing / Bakeout Cycles
- Pump Performance
- Chamber contaminants

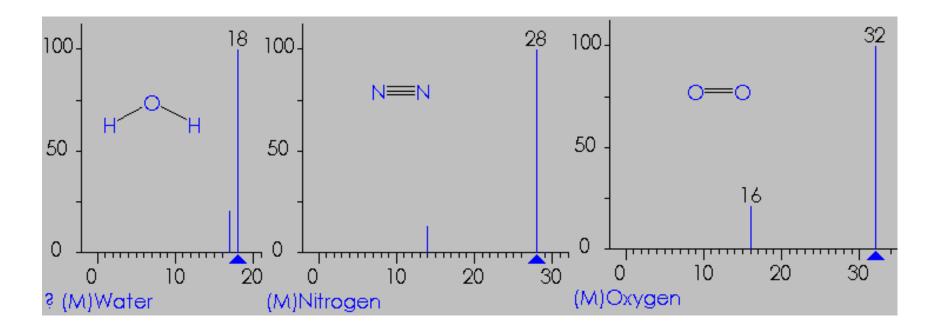


• Characterise your system and process for optimum results



Typical contaminant species present may be readily identified:

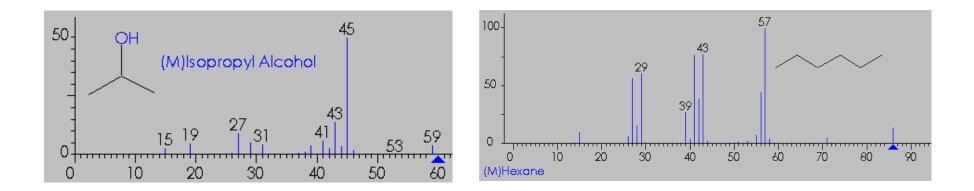
- Air leak: m/e 28 / 32 (ca. 4.5:1 ratio) confirm by the presence of peaks at m/e 14, 16
- Water: m/e 18 confirm by m/e 17





Typical contaminant species present may be readily identified:

 Hydrocarbons: characteristic groups of peaks, typical peaks at m/e 57, 55, 45, 43
 High mass peaks - back-streaming of oil or Vacuum Grease
 Low mass peaks - Cleaning fluid / solvent residue





Residual Gas Analysis : How it works

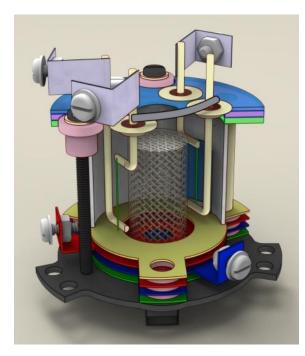
IONISATION – Electron Impact Ionisation (EI)

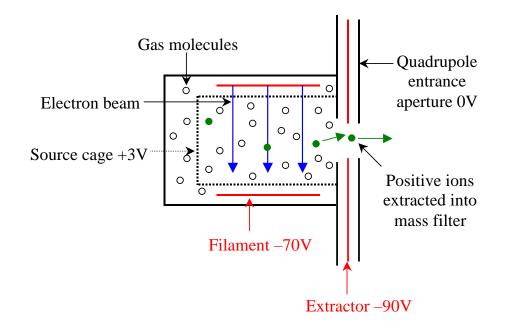
- Operation depends on the conversion of gas molecules into charged particles, typically positive ions / fragments.
- Achieved by electron impact ionisation via thermionic emission from a hot filament.
- A typical current is 1x10⁻⁴ Amps.
- Ions extracted into the mass filter.
- *Note:* Ionisation depends on the nature of the species involved.
- If a species is readily ionised it produces a higher MS signal than one which is poorly ionised.
- Use N₂ as a standard, RELATIVE SENSIVITY = 1
- *c.f.* Benzene = 5.9 and Helium = 0.14.



Residual Gas Analysis: How it works

IONISATION – Electron Impact Ionisation (EI)







How it works : IONISATION

The choice of filament material is important:

| Material | Operating Temperature | Comments | | |
|---------------|--------------------------|--|--|--|
| Yttria coated | 1900K | Good general purpose, robust. | | |
| Iridium | 1900K | Hiden standard. | | |
| | | Too hot for RGA use. Reacts with | | |
| Tungsten | 2400K | oxygen to give CO, CO2. Hiden uses | | |
| | | for certain applications. | | |
| Rhenium | 2300K | Too hot for RGA use. Reacts to form insulator coating. Hiden do not use. | | |
| Lanthanum | 1300K | Too brittle for RGA use. | | |
| hexaboride | | Hiden do not use. | | |

Lower operating temperature = lower outgassing



How it works : IONISATION

It is also important to note that EI yields several types of ions:

| Ion | Description | | |
|----------------|---|--|--|
| Molecular | The molecule with a positive charge by loss of an electron | | |
| Base | The most abundant ion in the spectrum | | |
| Fragment | Formed by cleavage of one or more bonds in the molecule | | |
| Rearrangement | Formed by bond cleavage and atomic migration | | |
| Doubly Charged | 2+ Ions from 2 electron loss steps appearing at 1/2 mass i.e. m/2 | | |
| Metastable | Fragmentation of ion into an ion of lower mass + a neutral particle | | |

See cracking pattern section for further details. Note: Fragment ions are also known as Product or Daughter Ions



How it works : THE MASS FILTER

- The mass filter differentiates the ions produced and selects species for detection.
- The most common form of mass filter is the Quadrupole.
- A Quadrupole is 2 pairs of parallel, equidistant metal rods (poles) biased at equal, but opposite potentials
- These twin potentials contain fixed DC and alternating RF components. By varying the RF component the resultant field produced by the rods may be varied.



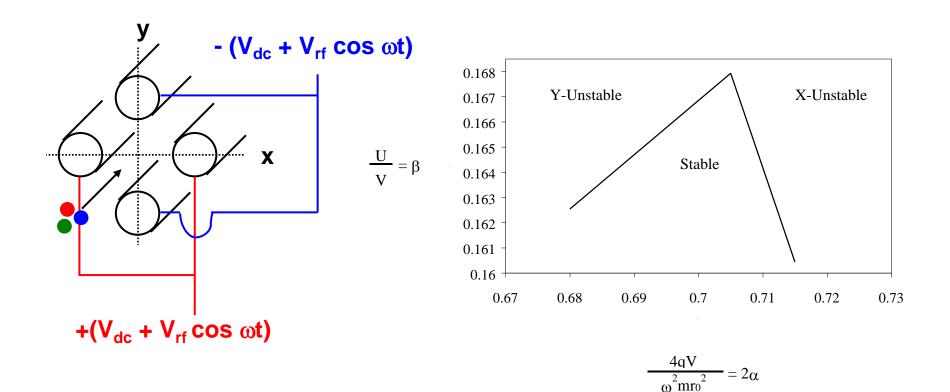


How it works: THE MASS FILTER

- Any ions entering the quadrupole field experience potential differences deflecting them from their original trajectory.
- The extent of deflection of any ion entering the field is related to its mass : charge (m/e or m/z) ratio.
- At each interval on the RF scan only one m/e ratio resonates with the field allowing the ion to pass along the z-axis.
- All other species are deflected and neutralised by impact upon the rods of the quadrupole.

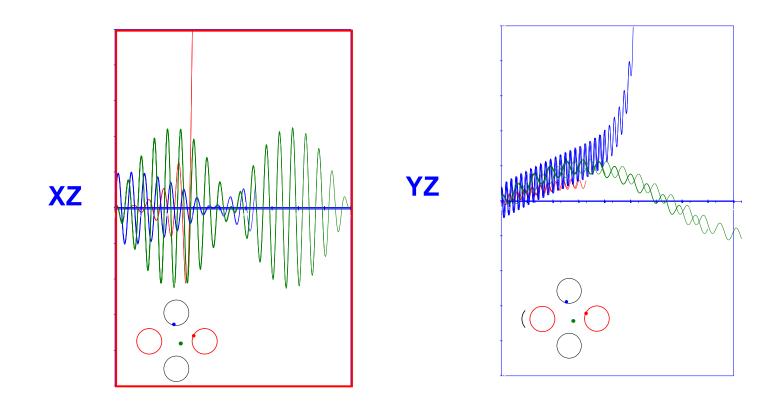


How it works : THE MASS FILTER





How it works : THE MASS FILTER -Mathieu Stability Diagrams





How it works : THE DETECTOR

• Filtered ions strike the detector to result in an ion current which is measured by a sensitive amplifier.

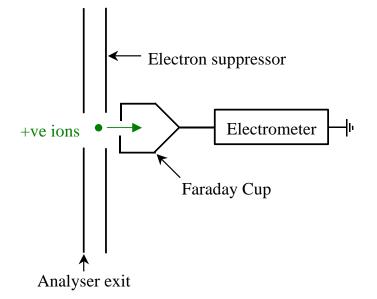
Two main types of Detector:

- a) The Faraday Cup an earthed passive conducting surface with a suppressor electrode to avoid false measurement.
- Fast moving ions strike the cup cause a 'shower' of 'secondary' electrons. The use of the 'cup' rather than a plate, allows all electrons to be collected.
- Hence, one ion arriving at the Faraday needs one electron for neutralisation but causes several electrons to be emitted; this provides amplification – several electrons for each ion.



How it works : THE DETECTOR

- The Faraday Cup: Detection limits
- Ion current for N₂ is 10⁻⁴ amps / mbar
- At 10⁻⁸ mbar of N₂, 10⁻⁸ * 10⁻⁴ = 10⁻¹² amps
- At 10^{-11} mbar of N₂ = 10^{-15} amps



 \Rightarrow Detection limit for conventional analogue amplifier



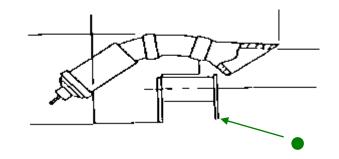
How it works : THE DETECTOR

• **b) Secondary Electron Multiplier** / Single Channel Electron Multiplier (SEM / SCEM):

A surface designed to generate secondary electrons.

- The ion impacts the surface generating 2 or 3 electrons each of which undergo further surface collisions generating more electrons, and so on in a cascade effect.
- Power for this cascade provided by an applied voltage.
- Gain is typically 10³ (10² for a channel plate)

Minimum detectable pressure 10⁻¹⁴ mbar to 10⁻¹³ mbar





Detector Pros and Cons:

Faraday Cup :

- Lower cost
- Indestructible
- Accurate
- BUT:
- Detection limit 10⁻¹¹ mbar
- Measurement relatively slow near detection limit



SEM / SCEM :

- Detection limit 10⁻¹⁴ 10⁻¹³ mbar with analogue detection
- Faster measurement

BUT :

- Expensive/Expendable/Sensitivity species dependent
- Sensitivity time / application dependent
- Typical maximum pressure of 10⁻⁵ 10⁻⁶ mbar





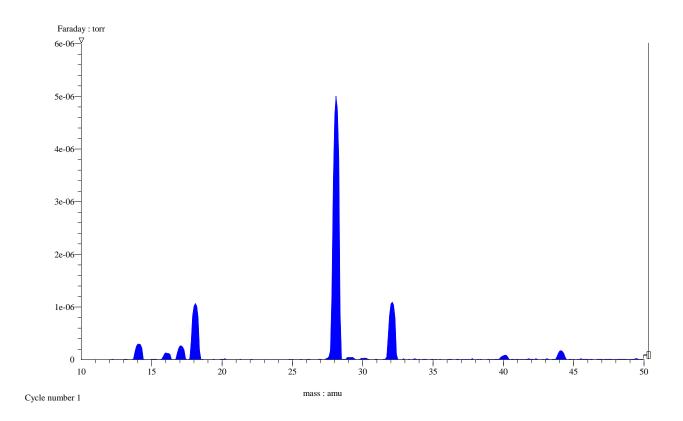
Resolution :

- The ability to separate /resolve ions of different m/e ratios
- All definitions directly / indirectly relate peak width to height
- For example:
 M / ΔM
- M at 10% peak height where $\Delta M < 1$ a.m.u.
- *i.e.* For any given mass M, the peak width at 10% of the peak height, measured from the baseline, is less than 1 amu
- This may be complicated if the mass peaks of trace species occur in the peak tail of a major species e.g. the detection of m/e 27 or m/e 29 in the presence of N₂ at m/e 28.



Interpretation and Cracking Patterns:

RGA data can be presented as a profile of mass / charge peaks.
 e.g. the RGA of Air :





Cracking Patterns:

 Cracking arises during ionisation when the high energy electrons used not only ionise species but fragment them. For CO: ¹²C¹⁶O + e⁻ → (¹²C¹⁶O)⁺ lonisation to give a peak at m/e = 28 ¹²C¹⁶O + e⁻ → ¹²C + ¹⁶O⁺ Cracking to give a peak at m/e = 16 ¹²C¹⁶O + e⁻ → ¹²C⁺ + ¹⁶O Cracking to give a peak at m/e = 12
 This fragmentation can be used to differentiate isobaric species: ¹²C¹⁶O from ¹⁴N₂ for example.

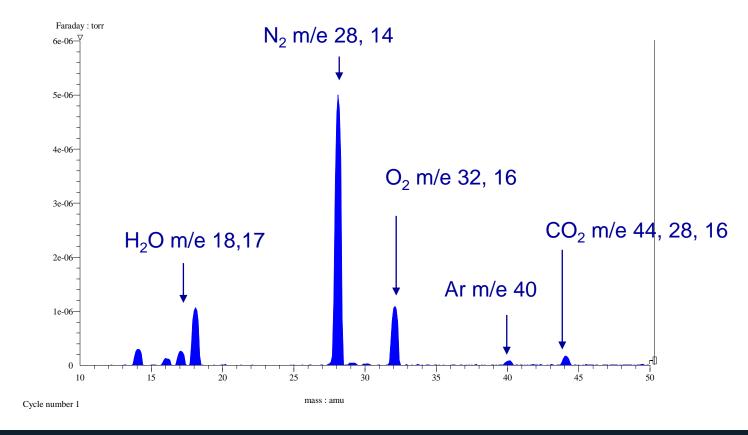
 $^{14}N_2$ has peaks at m/e 28 ($^{14}N_2^+$) and m/e 14 ($^{14}N^+$) from:

 $\label{eq:scalar} \begin{array}{ll} {}^{14}\,\mathsf{N}_2 + \,\mathrm{e}^{\scriptscriptstyle -} \rightarrow ({}^{14}\mathbf{N}_2)^{\,\ast} & \text{lonisation to give a peak at m/e = 28} \\ {}^{14}\,\mathsf{N}_2 + \,\mathrm{e}^{\scriptscriptstyle -} \rightarrow {}^{14}\,\mathsf{N} + {}^{14}\,\mathsf{N}^{\,\ast} & \text{Cracking to give a peak at m/e = 14} \\ ({}^{14}\,\mathsf{N}_2)^{\,\ast} + \,\mathrm{e}^{\scriptscriptstyle -} \rightarrow ({}^{14}\,\mathsf{N}_2\,)^{\ast\ast} & \text{lonisation to give a peak at m/e = 14} \\ \end{array}$ Note: CO/N₂ are ISOBARIC \Rightarrow the same mass but different composition



Interpretation and Cracking Patterns:

- Using Cracking Patterns it is possible to identify all species
- NOTE: The cracking pattern is directly related to the energy of the electrons used i.e. Under normal conditions the Cracking Pattern is characteristic of a species





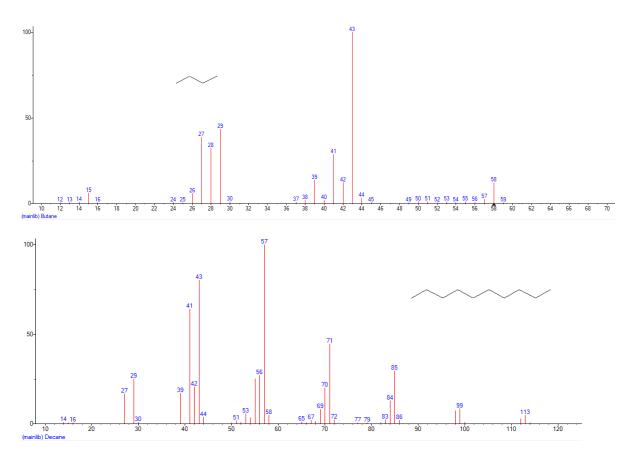
Interpretation and Cracking Patterns:

- Cracking pattern example Linear Hydrocarbons
- Decane ($C_{10}H_{22}$) and Butane (C_4H_{10}) show similarities why?

Both compounds show Clusters of Peaks at:

> m/z 57 m/z 43 m/z 29

 \Rightarrow Loss of unit of mass 14





Interpretation and Cracking Patterns:

Cracking pattern example The mass 14 unit \Rightarrow loss of alkyl-type species CH₂ i.e.

 $C_{10}H_{22}+e^{-} \rightarrow C_{10}H_{21}^{+} +e^{-} \rightarrow CH_{3}^{+}, C_{2}H_{5}^{+}, C_{3}H_{7}^{+}, C_{4}H_{9}^{+}, C_{5}H_{11}^{+} etc.$

And

 $C_4H_{10}+e^{\scriptscriptstyle -}\rightarrow C_4H_9^{\scriptscriptstyle +}+e^{\scriptscriptstyle -}\rightarrow CH_3^{\scriptscriptstyle +},\,C_2H_5^{\scriptscriptstyle +},\,C_3H_7^{\scriptscriptstyle +}$

 \Rightarrow FRAGMENTATION – Bond cleavage AND Ionisation.



Interpretation:

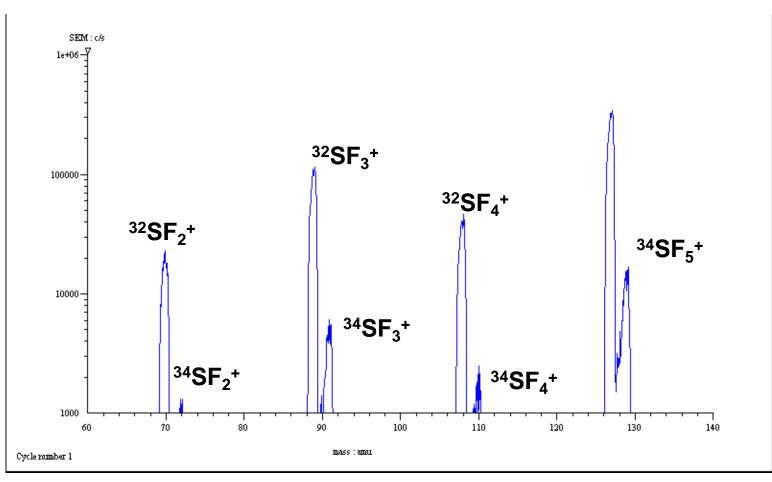
- Isotopic Abundance.
- Many species exist as several naturally occurring isotopes:

| Atom | Isotopes and relative abundances | | |
|-------------|--|--|--|
| Carbon | $^{12}C - 100, ^{13}C - 1.1$ | | |
| Nitrogen | 14 N $- 100, ^{15}$ N $- 0.4$ | | |
| Oxygen | 16 O - 100, 18 O - 0.2 | | |
| Fluorine | Monoisotopic | | |
| Chlorine | $^{35}\text{Cl} - 100, ^{37}\text{Cl} - 32.5$ | | |
| Bromine | 79 Br – 100, 81 Br - 98 | | |
| Iodine | Monoisotopic | | |
| Phosphorous | Monoisotopic | | |
| Sulfur | 32 S - 100, 34 S - 4.4 | | |
| Silicon | 28 Si - 100, 29 Si - 5.1, 30 Si - 3.4 | | |



Interpretation :

• Isotopic Abundance *e.g.*: Sulphur isotopes in the MS of SF₆



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Interpretation:

• General steps in interpretation of a Mass Spectrum.

1. Look for the Molecular Ion(s)

- 2. Note the general appearance of the spectrum
- 3. Check spectrum for peak clusters for Isotope patterns
- 4. Check for low-mass neutral fragment loss e.g. CH₂
- 5. Check for characteristic low-mass fragments
- 6. Compare to reference spectra



Quadrupole Mass Spectrometers

Hiden manufactures a wide range of MS systems, all tailored to specific customer applications.

Options include:

- Mass Ranges of 50, 100, 200, 300, 510, 1000 and 2500 amu
- Ioniser options including cross-beam, gold plated and platinum
- Detectors: Faraday and electron multipliers, Channelplate, Channeltron, analogue current measurement and digital pulse ion counting options
- 6 mm, 9 mm or 20 mm pole diameter
- Single and triple filter options
- Ethernet, USB, and serial comms.





Hiden Gas Analysers

- Rugged modular construction with precision machined radial ceramic rod supports
- Powerful processor with data buffering for true multi-tasking operation
- Ion blast free for maximum sensitivity in He leak detection
- Detection to 5 x 10⁻¹⁴ Torr / PPB levels
- Bench, cart or console mounted
- Fully automated operation
- Application specific gas inlets
- Corrosive gas / oil free pumping
- Multi-stream options
- Quantitative gas analysis





Appendix 1

• Table 1a gives some of the common RGA contaminants

| Name | Formula | Peak 1 m/e % | Peak 2 m/e % | Peak 3 m/e % | rel sens |
|----------------------|---------------------------------|-----------------|-----------------|-----------------|-------------|
| Acetone | C ₃ H ₆ O | 43 100 | 58 33 | 15 20 | 3.6 |
| Air | | 28 100 | 32 27 | 14 6 | 1.0 |
| Ammonia | NH ₃ | 17 100 | 16 80 | 15 8 | 1.3 |
| Argon | Ar | 40 100 | 20 16 | 36 0.3 | 1.2 |
| Benzene | C_6H_6 | 78 100 | 77 19 | 52 16 | 5.9 |
| Boron Trichloride | BCl ₃ | 81 100 | 58 33 | 15 20 | 1.0 |
| Carbon Dioxide | CO ₂ | 44 100 | 16 9 | 14 6 | 1.4 |
| Carbon Monoxide | СО | 28 100 | 12 5 | 16 2 | 1.05 |
| Carbon Tetrafluoride | CCl ₄ | 69 100 | 50 12 | 19 7 | 1.0 |
| Diborane | B_2H_6 | 26 100 | 27 97 | 24 90 | 1.0 |
| Ethane | C_2H_6 | 28 100 | 27 33 | 30 26 | 2.6 |
| Fomblin Oil | | 69 100 | 20 28 | 16 16 | 1.0 |
| Freon 12 | CCl_2F_2 | 85 100 | 87 32 | 50 16 | 2.7 |
| Helium | He | 4 100 | | | 0.14 |
| Hydrogen | H_2 | 2 100 | 1 2 | | 0.44 |



Appendix 2

 Table 1b gives more common RGA contaminants

| Name | Formula | Peak 1 m/e % | Peak 2 m/e % | Peak 3 m/e % | rel sens |
|-----------------------|--------------------|-----------------|-----------------|-----------------|-------------|
| Hydrogen Chloride | HCl | 36 100 | 38 32 | 35 17 | 1.6 |
| Hydrogen Sulfide | H_2S | 34 100 | 32 44 | 33 42 | 2.2 |
| Krypton | Kr | 84 100 | 86 31 | 82 21 | 1.7 |
| Methane | CH ₄ | 16 100 | 15 85 | 14 16 | 1.6 |
| Methanol | CH ₃ OH | 31 100 | 32 67 | 29 65 | 1.8 |
| Neon | Ne | 20 100 | 22 10 | 21 0.3 | 0.23 |
| Nitrogen | N_2 | 28 100 | 14 5 | 29 1 | 1.0 |
| Oxygen | O_2 | 32 100 | 16 9 | | 0.86 |
| Phosphine | PH ₃ | 34 100 | 33 33 | 31 32 | 2.6 |
| Pump Oil | | 57 100 | 55 73 | 43 73 | 1.0 |
| Silane | SiH4 | 30 100 | 31 78 | 29 29 | 1.0 |
| Silicon Tetrafluoride | SiF ₄ | 85 100 | 86 5 | 28 4 | 1.0 |
| Sulfur Dioxide | SO ₂ | 64 100 | 48 50 | 32 10 | 2.1 |
| Water | H ₂ O | 18 100 | 17 21 | 16 2 | 0.9 |
| Xenon | Xe | 132 100 | 129 98 | 131 79 | 3.0 |



- www.HidenAnalytical.com
- The Hiden website is an excellent resource with product pages, brochures, catalogues, product pages with some application notes, presentation and other information.
- Contact +44 1925 445225 for direct support.

