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Gas Analysis | DEMS Technical Information TI-20022.2

## **Summary of DEMS / OEMS Options**

Differential Electrochemical Mass Spectrometry (DEMS) and Online Electrochemical Mass Spectrometry (OEMS) require a diverse range of inlet options to allow analysis of both dissolved and evolved gases.

Measurement of dissolved gases is typically performed using a PTFE membrane which allows dissolved gas products to pass to the MS via a differentially pumped vacuum system.

Measurement of evolved gases is by direct analysis of product gas in a headspace with or without carrier gas flow.

This document will outline the applicability of each inlet to DEMS and OEMS applications.

### **Hiden Analytical DEMS Cells**

### Type A cell



Figure 1: Type A Cell Schematic

The Type A DEMS cell is a thin layer cell, particularly appropriate for the study of desorption products under static electrolyte conditions but is applicable to many types of electrochemistry utilising an aqueous electrolyte [1].

If flow is continuous then it must be low enough that the products are not removed from the cell before crossing the thin layer of



electrolyte to the MS, it is recommended to use a syringe pump delivering  $1\mu$ /s.

The cell includes a 5mm diameter vitreous carbon working electrode and 1/16" Ag/AgCl reference electrode.

Counter electrode is not supplied and can either enter the cell directly or be in contact with electrolyte outside of the cell.

Materials can be deposited onto the face of the working electrode. The electrolyte and counter and reference electrodes enter the cell through 1/16" holes.

The PEEK body and PTFE membrane allows for use of corrosive electrolytes.

| Suitable for         | Not suitable for      |
|----------------------|-----------------------|
| Dissolved gas        | Evolved gas           |
| Static conditions    | High flow             |
| Aqueous electrolytes | Volatile electrolytes |
| Vitreous Carbon WE   |                       |

### Type B cell





The Type B DEMS cell is a dual thin-layer cell, most suitable for monitoring continuous faradaic reactions, with controlled hydrodynamics in the determination of product formation rates and turn over frequencies [1].

Particularly suitable for CO<sub>2</sub> Reduction [2], the Type B cell employs dual thin layer design to compartmentalise the anode and cathode chambers using an ion conducting membrane.

The cell employs a parallel electrode configuration, high surface area electrodes and low volume catholyte layer. This gives a low cell resistance ( $\sim$ 50  $\Omega$ ), robust electrode connectivity, and minimal overpotential at the counter electrode, enabling the potentials required to produce



hydrocarbons and alcohols over polycrystalline copper to be experimentally accessible.



Figure 3: Type Schematic

The cell requires a low flow of electrolyte through the 2 compartments with only products formed in the cathode compartment measured by the MS. Recommended flow is 2ml/min delivered using a dual syringe pump.

Supplied with platinum counter electrode and Ag/AgCl reference electrode with options of solid Cu / Ag / Au plated or vitreous carbon working electrode.

The Type B cell is supplied with clear acrylic anode chamber to allow user assessment of bubble production but also supplied with PEEK anode chamber for corrosive electrolytes. Cathode chamber uses a PEEK body.

| Suitable for         | Not suitable for      |
|----------------------|-----------------------|
| Dissolved gas        | Evolved gas           |
| Low flow conditions  | High flow             |
| Aqueous electrolytes | Volatile electrolytes |
| Cu, Ag, Au, Vitreous |                       |
| Carbon WE            |                       |



### **Type B Modified Cathode Chamber**

The modified cathode chamber allows for alternative working electrode materials by direct sputtering on to the membrane surface [3].



Figure 4: Direct sputtering of membrane surface

Coating the membrane surface, allows volatile species at the electrode-electrolyte interface to be sampled. Furthermore, the delay time between product generation and detection is minimized and the liquid-phase product collection efficiency is maximized.



Figure 5: Modified Type B Cathode Chamber

Larger chamber volume is suitable for higher flow rates through the cell, avoiding bubble hold up.

The modified chamber utilises the same anode chamber used in the standard Type B cell.



Figure 6: Schematic of Modified Type B Setup

| Suitable for         | Not suitable for      |
|----------------------|-----------------------|
| Dissolved gas        | Evolved gas           |
| High flow conditions | Volatile electrolytes |
| Aqueous electrolytes |                       |
| Sputtered WE         |                       |



### **DEMS Probe**



Figure 7: DEMS 6mm Sampling Probe

The DEMS probe allows dissolved gas to be measured in custom cells. The 6mm probe sits at the end of 1/16" flexible stainless-steel tubing so can be moved between experiments easily.

It utilises the same PTFE membrane and differential pumping techniques of the DEMS cells. Changes to the instrument by-pass pumping and to the membrane interface are employed to optimise the response of the mass spectrometer to the remote probe design. A bypass pumping restrictor is employed and thicker PTFE membrane (50 microns) is used to achieve the optimum MS sampling pressure for maximum sensitivity and fast response time.

Response time is typically < 2s.

The probe should be positioned as close as possible to the WE.

The 1/16" probe tube diameter and 6mm probe is particularly suitable for a range of Redox.me cells.



Figure 8: DEMS probe fitted inside Redox.me cell

| Suitable for         | Not suitable for      |
|----------------------|-----------------------|
| Dissolved gas        | Evolved gas           |
| Static or flowing    | Low electrolyte       |
| conditions           | volume                |
| Aqueous electrolytes | Volatile electrolytes |
| Redox.me cells       |                       |
| Custom cells         |                       |



# Inlets for OEMS (Online Electrochemical MS)

### **Ultra-Low Flow Capillary**

The Ultra-low flow Capillary allows measurement of evolved gas with minimal dilution of the product gas by carrier gas flow. At 200µl/min, flow is similar to that recommended [4] for use with the ECC-DEMS cell and PAT cell from EL-Cell but is also suitable for connection to Redox.me or a custom cell.



Figure 9: Ultra Low Flow capillary connected to Redox.me cell

Corrosive resistant options make the capillary suitable for Lithium battery testing and use of volatile electrolytes.

The recommended experimental set up would allow a controlled flow of carrier gas to flow past the inlet of the cell whilst the capillary is connected at the cell outlet. This ensures carrier gas flow is set at 200µl/min.

A custom quick connect kit for the ECC-DEMS cell, allows for oxygen free cell loading and subsequent analysis, outside of an Argon box.



Figure 10: ULF connection to ECC-DEMS cell, with quick connects

The heated QIC inlet gives fast response and ensures condensable gases remain as vapours during sampling.

| Suitable for          | Not suitable for     |
|-----------------------|----------------------|
| Evolved gas           | Dissolved gas        |
| Low product gas flow  | No carrier gas flow  |
| Condensable vapours   | Very low product gas |
|                       | flow                 |
| Fast response         | Multiple streams     |
| Various cells         |                      |
| Quantitative analysis |                      |
| Volatile electrolytes |                      |



### **Microflow Capillary**



Figure 11: Single Microflow Capillary

For applications with only very low levels of product gas, where dilution with carrier gas may decrease concentrations below the limit of detection, the micro-flow capillary is recommended.

With flow rates of either 12 or 25µl/min, headspace gas can be sampled without significantly changing pressure in the cell.

Connection is via 1/16" fitting so is easily fitted to either custom or commercial cells.

Capillary length is either 1m (25µl/min) or 2m (12µl/min), giving response times of 8s and 16s respectively (for air). A filter helps avoid particulates blocking the capillary but because it is unheated the possibility of condensation is increased.

Option for up to 8 multiplexed capillaries to allow automated sequential sampling, from multiple sample points or cells.



Figure 11: 8-way option for Microflow capillary

| Suitable for          | Not suitable for  |
|-----------------------|-------------------|
| Evolved gas           | Dissolved gas     |
| Very Low gas flow     | Fast response     |
| Multiple streams      | Condensable gases |
| Various cells         |                   |
| Quantitative analysis |                   |
| Volatile electrolytes |                   |



### **Standard QIC Capillary**

When cell volume or product gas levels are high then the standard QIC capillary can be used.

Flow rates from 800µl to 28ml/min are available giving fast response time (300ms) along the heated capillary. Increased capillary length (2m) gives more flexibility.

The capillary can be connected to a multistream valve so that up to 16 streams can be measured using a single capillary.

The QIC capillary flow rate can be changed easily by swapping the capillary liner and so provides a versatile solution for many applications as well as OEMS.

| Suitable for          | Not suitable for    |
|-----------------------|---------------------|
| Evolved gas           | Dissolved gas       |
| High gas flow         | No carrier gas flow |
| Multiple streams      | Small cell volume   |
| Various cells         |                     |
| Quantitative analysis |                     |
| Volatile electrolytes |                     |
| Fast Response         |                     |
| Versatility           |                     |
| Condensable gases     |                     |

### Summary

The DEMS/OEMS options detailed in this TI sheet are user interchangeable. The options available illustrate the versatility of the Hiden HPR-40 DEMS mass spectrometer system for DEMS/OEMS and other real time gas/vapour analysis applications.

In addition to the many applications in electrochemistry illustrated in this TI sheet, sample inlets are available for MIMS, TG-MS and other applications where real time analysis of gases and vapours is required.

#### References

- 1. Ashton S J (2012), Design, Construction and Research Application of a Differential Electrochemical Mass Spectrometer (DEMS), Springer Theses
- Clark, Ezra L., et al. (2015) Differential electrochemical mass spectrometer cell design for online quantification of products produced during electrochemical reduction of CO2. Analytical chemistry 87.15: 8013-8020.
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- 4. Z. Peng at al. (2012) A Reversible and Higher-Rate Li-O<sub>2</sub> Battery, Science, Vol. 337 no. 6094 pp. 563-566